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Adrian has 30 years of
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KEY page 4 MM = $(4.68 \times 0.60) / (0.5 \times 0.1)$...

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(Form B) ... Total Score

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10 points 2. Answer the following questions that relate to chemical reactions. (a) Iron(III) oxide can be reduced with carbon monoxide according to the following equation.

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The AP Chemistry Exam covers a full-year introductory college course in chemistry with laboratory. The

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multiple-choice questions in Section I cover the breadth of the curriculum. Section II, the free-response part of the exam, includes three quantitative problems, one question on writing chemical reactions and predicting products, and two essays.

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Board Subject: AP

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page 2 1993 I. $2 \text{Mn}^{2+} + 4 \text{OH}^- + \text{O}_2(\text{g}) \rightarrow 2 \text{MnO}_2(\text{s}) + 2 \text{H}_2\text{O}$ II.

$\text{MnO}_2(\text{s}) + 2 \text{I}^- + 4 \text{H}^+ \rightarrow \text{Mn}^{2+} + \text{I}_2(\text{aq}) + 2 \text{H}_2\text{O}$ III. $2 \text{S}_2\text{O}_3^{2-} + \text{I}_2(\text{aq}) \rightarrow \text{S}_4\text{O}_6^{2-} + 2 \text{I}^-$

The amount of oxygen, O_2 , dissolved in water can be determined by

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titration. First, MnSO_4
and NaOH

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[delta] and [sigma] are
used to indicate the
capital Greek letters.
[square root] applies to
the numbers enclosed
in parenthesis
immediately following

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Using the information in the table above, answer the following questions about organic compounds. (a) For propanone, (i) draw the complete structural formula (showing all atoms and bonds);

**AP* Bonding &
Molecular Structure
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$$K_{sp} = [\text{Sr}^{2+}][\text{F}^-]^2 = 7.9 \times 10^{-10} = (7.6 \times 10^{-6})$$

$$(z)^2 = 7.9 \times 10^{-10} \quad z =$$

$$1.0 \times 10^{-2} \text{M } \% \text{F}^- \text{ still in}$$

$$\text{solution} = 1.0 \times 10^{-2} /$$

$$2.0 \times 10^{-2} \times 100 =$$

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50.% Page 2. Page 3
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[delta] and [sigma] are
used to indicate the
capital Greek letters.
[square root] applies to
the numbers enclosed
in parenthesis
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assumptions are
justified within 5%.

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