

## Chapter 12 Lecture Notes Carbohydrates

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carbohydrates-that is, their considerable solubility in water, instability to strong oxidizing agents and acidic or basic reagents, reluctance to crystallize, and their tendency to decompose rather than give sharp melting points.

### CARBOHYDRATES

• Carbohydrates are grouped into 3 classes: -Monosaccharides are the simplest sugars and serve as the building blocks of larger molecules Example: Glucose -Oligosaccharides contain 2-10 monosaccharides bonded together (building block = residue) Example: Sucrose -Polysaccharides contain more than 10 residues Example: Complex Carbohydrates

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All carbohydrates are hydrates of carbon and they contain C, H and O. The ratio of hydrogen and oxygen in the majority of carbohydrates will be in 2:1 as in water. Some carbohydrates also contain nitrogen, phosphorous and sulfur. Majority of carbohydrates, not all, have the empirical formula  $(CH_2O)_n$ . In biochemistry, carbohydrates are denoted as saccharides.

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- Carbohydrates are compounds of tremendous biological importance: - they provide energy through oxidation - they supply carbon for the synthesis of cell components - they serve as a form of stored chemical energy - they form part of the structures of some cells and tissues
- Carbohydrates, along with lipids, proteins, nucleic acids, and other compounds are known as

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