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Exercise Solutions

R13.1 Recursion

Recursion solves a problem by using the solution of the same problem with simpler values. Iteration

Iteration uses some form of a loop (for loop, while loop) to solve a problem. Infinite recursion Infinite

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recursion occurs when the recursion never stops.

## **Chapter 13 Review Solutions - Chapter 13 Review Exercise**

...

CHAPTER 13 REVIEW  
Solutions SECTION  
13-2 SHORT ANSWER  
Answer the following questions in the space provided. 1. Following are statements about the dissolving process. Explain each one at the

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molecular level. a.

Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.  
b.

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ANSWER KEY 12. A  
battery is like a water  
pump because the  
battery supplies  
electrical potential  
energy to a circuit and  
a pump provides  
potential energy to  
water. 13. Multimeters



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can measure both voltage and current for AC and DC while ammeters measure only current and voltmeters only voltage (potential difference).

14.

## **Chapter 13 Review**

### **Answer Key**

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Solutions Section 1

Answers Chapter 13  
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Section Yeah,  
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Chemistry Chapter 13

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review. absolute zero.  
universal gas constant.  
Boyle's law. Charles's  
law. a temperature at  
which a gas  
theoretically occupies  
zero volume. 0.08206 L  
atm/mol K. at constant  
temperature, the  
volume of a given  
amount of gas i.... at  
constant pressure, the  
volume of a given  
amount of gas is d....

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**Flashcards and  
Study Sets ...**

CHAPTER 13 REVIEW

Ions in Aqueous

Solutions and

Colligative Properties

MIXED REVIEW SHORT

ANSWER Answer the

following questions in

the space provided. 1.

Match the four

compounds on the

right to their

descriptions on the left.

b an ionic compound

that is quite soluble in

water (a) HCl c an ionic

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compound that is not  
very soluble in water  
(b)  $\text{NaNO}_3$

## **13 Ions in Aqueous Solutions and Colligative Properties**

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- General Chem ...**

Chapter 13 Review

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Problems

INSTRUCTIONS: Write your answers on lined or blank sheets of paper. You do not need to write the question, ONLY WRITE THE PROBLEM NUMBER and ANSWERS/SOLUTIONS.

- For problems that involve calculations, you must show your work to get full credit.

**Chapter 13 Review Problems -**

*Page 14/24*

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Chapter 13:

Electromagnetic

Induction 13-5

Statement: The current  
in the secondary circuit

is 12 A. 34. Solve for I

in the power equation

$P = VI$  and then

substitute the values

given for P and V to

find I, which is the

amount of current

produced:  $P=VI$   $I= P/V$

$P = 1500 \text{ MW}; V = 20.0$

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$$kV I = P V = 1500MW$$

$$20.0kV = 1.5 \times 10^9 W$$

$$2.0 \times 10^4 V$$

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Chapter 13 Review  
Answer Key

Understanding

Vocabulary Section

13.1 1. electric current

2. electrical symbols 3.

switch 4. open circuit

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5. closed circuit 6.  
resistance Section 13.2  
7. ampere 8. battery 9.  
voltage 10. volt Section  
13.3 11. ohm 12.

**Chapter 13 Review  
Solutions Answer  
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CHAPTER 12 REVIEW  
Solutions SECTION 2  
SHORT ANSWER

Answer the following  
questions in the space  
provided. 1. The  
following are  
statements about the

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dissolving process.

Explain each one at the molecular level. a.

Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

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Solutions for Problems in Chapter 13 is solved.

1CE; 1P; 1Q; 2CE; 2P;  
2Q; 3CE; 3P; 3Q; 4CE;  
4P; 4Q; 5CE; 5P; 5Q;  
6CE; 6P; 7CE; 7P; 8CE;

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9CE; 10CE; 11CE;  
12CE; 13CE; 14CE

**Chapter 13 Solutions**  
**| Introductory**  
**Econometrics 4th ...**

CHAPTER 12 REVIEW  
Solutions MIXED  
REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1. Solid  $\text{CaCl}_2$  does not conduct electricity. Explain why it is considered to be an electrolyte.  $\text{CaCl}_2$  is

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an ionic solid. In the crystal form, its ions are locked in position.

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Chapter Thirteen: Ions in Aqueous Solutions and Colligative Properties. Section 1: Chapter review 8 thru 13. Section 2: Chapter review 19 thru 34 (odd). Boiling point elevation and freezing point depression:

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Raising or lowering the boiling or freezing point of a solution by adding solute.

### **Chapter Thirteen** **[Ions in Aqueous** **Solutions and ...**

Chapter 13 Review Solutions Answer Key. starting the chapter 13 review solutions answer key to retrieve all daylight is satisfactory for many people. However, there are yet many people

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who after that don't  
when reading. This is a  
problem. Chapter 13  
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Recursion

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